

COBAMIDE-DEPENDENT SYNTHESIS OF METHIONINE: LIGHT

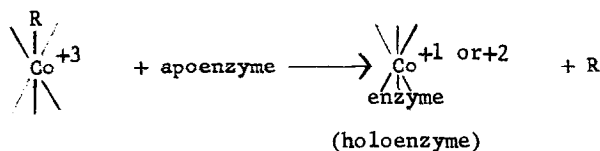
REACTIVATION OF AN INHIBITED ENZYME

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It has been shown previously that ethyl-B₁₂¹ and β -propionate-B₁₂ can partially replace methyl-B₁₂ in the transfer of the methyl group from methyl-folate-H₄ to homocysteine to form methionine. Other alkyl cobamides tested (propyl-B₁₂, etc.) were not only inactive in the system, but also inhibitory (Weissbach, et al., 1964). Since a reduced cobamide has been postulated as the active cobamide (Weissbach, et al., 1963; Guest, et al., 1964; Kerwar, et al., 1964) in this reaction, it appeared that those alkyl-B₁₂ compounds active in the reaction combine with the apoenzyme, with a subsequent release of the alkyl moiety, leaving the active reduced cobamide species (Co⁺¹ or ⁺²) on the enzyme (Weissbach, et al., 1964).



The present communication examines in more detail the reaction of apoenzyme with alkyl-cobamide compounds.

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¹The following abbreviations will be used: methyl-B₁₂, 5,6-dimethylbenzimidazolylcobamide-methyl; ethyl-B₁₂, 5,6-dimethylbenzimidazolylcobamide-ethyl, etc.

Wild type Escherichia coli K₁₂, obtained from Dr. A. L. Taylor, was employed in the present studies. The organism was grown on a glucose-minimal media in the absence of any cobamide. The enzyme preparation used was a dialyzed ammonium sulfate fraction, (22 mg protein per ml) capable of forming 13 μ moles methionine per 30 min. per mg protein from methyl-folate-H₄ using previously reported conditions (Weissbach, et al., 1963; 1964). Since the organism was grown in the absence of any cobamide the enzyme present was in the form of apoenzyme.

Preincubations were performed as follows: 0.15 - 0.25 ml of enzyme, 20 μ moles KPO₄ buffer pH 7.4, and 1 μ mole of cobamide were incubated for 5 min. at 37° in a total volume of 0.3 ml. Three tenths ml of a charcoal suspension (5 mg) was added to stop the reaction and remove the excess cobamide. After centrifugation an aliquot of the supernatant fraction (0.04 - 0.08 ml) was assayed for both activity (Weissbach, et al., 1963) and percent holoenzyme present (see below). Activity is expressed as μ moles of methionine formed per 30 min. In some experiments (Table 3) an aliquot of the charcoal supernatant fraction was exposed (in ice) to a 200-watt bulb, at a distance of 15 cm for 5 min. before activity and percent holoenzyme were determined.

A convenient assay for holoenzyme formation was based on the cobamide requirements for the reaction with either apoenzyme or holoenzyme when the incubation was performed in air. With apoenzyme the cobamide serves two functions, to form holoenzyme as well as to lower the oxygen tension of the incubation by a nonspecific action, similar to that reported by Peel (1962) on the CO₂-pyruvate exchange reaction. Previous studies have shown that, under the conditions employed, methyl-B₁₂ (and to a lesser degree ethyl-B₁₂ and β -propionate-B₁₂) satisfied both cobamide requirements using apoenzyme, while other alkyl-B₁₂ compounds such as propyl-B₁₂, butyl-B₁₂ and deoxyadenosyl-B₁₂ did not (Weissbach, et al., 1964). In fact the latter derivatives have been shown to inhibit the enzymatic synthesis of methionine from N⁵-methyl-folate-H₄ in the presence of methyl-B₁₂ and apoenzyme. With

holoenzyme, which contains the active cobamide on the enzyme, a cobamide is also required in vitro for maximum activity under the aerobic conditions used; but only to satisfy the need for anaerobiosis. Thus, all the alkyl cobamide compounds tested fulfilled this function. By comparison of activity in the presence of methyl-B₁₂ and propyl-B₁₂ it was possible to assay the percentage of apo- or holoenzyme present. This is shown in Table 1. With apoenzyme no

TABLE 1
ASSAY FOR HOLOENZYME FORMATION

Cobamide added to reaction mixture ¹	Apoenzyme ² Activity	$\frac{\text{Propyl-B}_{12} \times 100}{\text{Methyl-B}_{12}}$	Holoenzyme ² Activity	$\frac{\text{Propyl-B}_{12} \times 100}{\text{Methyl-B}_{12}}$
Methyl-B ₁₂	1.9	<5	2.2	>90
Propyl-B ₁₂	<0.1		2.0	
None	<0.1		0.2 ³	

¹The reaction mixture contained in a total volume of 0.2 ml: homocysteine, 50 μ moles; S-adenosylmethionine, 10 μ moles; ATP, 100 μ moles; TPNH 100 μ moles, 2-mercaptoethanol, 40 μ moles, KPO₄ buffer, pH 7.4 10 μ moles; cobamide, 10 μ moles; DL-¹⁴C-methyl-folate-H₄, 30 μ moles and enzyme. The methionine formed was assayed as described previously (Weissbach, et al., 1963). Activity is expressed in terms of μ moles methionine formed per 30 minutes.

²Apoenzyme was prepared from cells grown without vitamin B₁₂, while holoenzyme was prepared from cells grown with the vitamin (0.01 μ moles/ml) present in the medium.

³Under the conditions used holoenzyme did not show an absolute cobamide requirement. Generally 10-30% of maximum activity observed in the presence of a cobamide was seen if no cobamide was present. This is very likely due to some degree of anaerobiosis in the incubations, since high levels of mercaptoethanol are employed. Under strictly anaerobic conditions no cobamide is needed with holoenzyme.

significant activity (<5%) is seen if propyl-B₁₂ is used in place of methyl-B₁₂ in the assay; while with holoenzyme propyl-B₁₂ can completely replace methyl-B₁₂. In the present study the results are expressed in percent holoenzyme, i.e., the percent of enzyme protein present as holoenzyme, as determined by the $\frac{\text{propyl-B}_{12}}{\text{methyl-B}_{12}}$ activity ratio.

The short-chain alkyl-B₁₂ compounds were synthesized by the method of Smith, et al (1962). Deoxyadenosyl-B₁₂ was a gift of Dr. D. Perlman, Squibb Institute. Large amounts of E. coli K₁₂ cells were kindly supplied by

Mr. D. Rogerson.

Preincubation of the enzyme with methyl-B₁₂ yielded excellent formation of holoenzyme. Ethyl-B₁₂ and β -propionate-B₁₂ were as active as methyl-B₁₂ in forming holoenzyme, agreeing with earlier results (Weissbach, et al., 1964) with these compounds. Other alkyl cobamides tested such as propyl-B₁₂, butyl-B₁₂ and deoxyadenosyl-B₁₂ did not yield significant formation of holoenzyme. It was also observed that preincubation with propyl-B₁₂, but not deoxyadenosyl-B₁₂, resulted in a marked inhibition (60-80%) of the enzyme. A typical experiment is shown in Table 2. No holoenzyme was formed with methyl-B₁₂ (Experiment 4, Table 2) when the preincubation was carried out in ice.

TABLE 2

ACTIVITY AND HOLOENZYME FORMATION AFTER PREINCUBATIONS OF APOENZYME WITH
ALKYL-COBAMIDE COMPOUNDS

Preincubation	Activity	% Holoenzyme
1. Enzyme + methyl-B ₁₂	4.0	53
2. Enzyme + propyl-B ₁₂	1.4	< 3
3. Enzyme + deoxyadenosyl-B ₁₂	4.6	< 3
4. Enzyme + methyl-B ₁₂ (0°)	3.8	< 3

Preincubations were done for 5 min. at 37° as described in text except for #4. Activity is expressed as μ moles methionine formed per 30 min. using equal aliquots from the preincubated tubes.

Enzyme, inactivated by preincubation with propyl-B₁₂, could be readily activated by exposing it to light. This treatment not only completely re-stored the enzymatic activity, but yielded holoenzyme, as shown in Table 3. No significant effect of light was observed on enzyme extracts preincubated with either methyl-B₁₂ or deoxyadenosyl-B₁₂. The latter alkyl derivative therefore neither forms holoenzyme nor an inhibited enzyme.

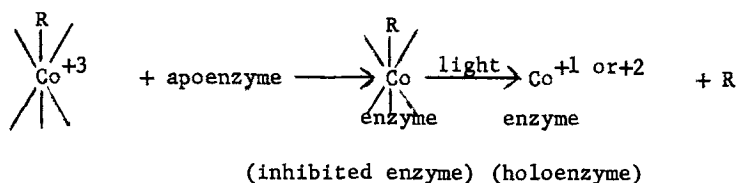
TABLE 3

EFFECT OF LIGHT ON ENZYME AFTER PREINCUBATION WITH ALKYL-COBAMIDE DERIVATIVES

Preincubation	Treatment	Activity	% Holoenzyme
Enzyme + methyl-B ₁₂	dark	5.3	62
	light	5.5	57
Enzyme + propyl-B ₁₂	dark	2.1	<3
	light	5.2	60
Enzyme + deoxyadenosyl-B ₁₂	dark	4.5	<3
	light	4.5	<3

Preincubation conditions and light treatment are described in the text. Activity is expressed as μ moles methionine formed per 30'.

The above experiments support the view that cleavage of the carbon-cobalt bond is an essential step in the biological activity of alkyl cobamides in this reaction. The data suggest that the enzyme can break the carbon-cobalt bond in methyl-B₁₂, but cannot cleave the carbon-cobalt bond in propyl-B₁₂. Thus, reaction of apoenzyme with propyl-B₁₂ yields an inactive enzyme.



R = propyl

In this case, light accomplishes what the enzyme is unable to do, i.e., cleave the carbon-cobalt bond which is known to be light sensitive (Weissbach, et al., 1960; Dolphin, et al., 1962). The light cleavage of the carbon-cobalt bond has been shown to proceed through B_{12r} (Brady and Barker, 1961) which is further support for a role of a reduced cobamide species in this reaction.

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